

# Reactions of 3-substituted 1-aryl-5,6,7,8-tetrafluoroquinolones(cinnolones) with morpholine

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1-Aryl-3-ethoxalyl(heteryl)-5,6,7,8-tetrafluoroquinolones(cinnolinones) react with morpholine to give 7-morpholino- and 5,7-dimorpholino derivatives, depending on the reaction conditions.

**Key words:** 5,6,7,8-tetrafluoroquinolones, 5,6,7,8-tetrafluorocinnolones, aromatic nucleophilic substitution, morpholine.

Recently, a series of 1-aryl-3-ethoxalyl(heteryl)-5,6,7,8-tetrafluoro-1,4-dihydroquinolin(cinnolin)-4-ones (**1–5**) have been synthesized.<sup>1–3</sup> Like all organofluorine substances, these compounds enter into nucleophilic substitution reactions. Investigations of the behavior of structurally similar compounds, namely, derivatives of fluoroquinolone-3-carboxylic acids<sup>4</sup> and 5,6,7,8-tetrafluorochromen-4-one<sup>5</sup> showed that the monosubstitution reactions with alkylamines mainly give 7-substituted, less often 5-substituted, and, in specific cases, 8-substituted products. The formation of 5,7-disubstituted products was noted only once.<sup>4</sup>

In the present work, we studied the possibility of aromatic nucleophilic substitution in quinolones **1** and **3** and cinnolones **2**, **4**, and **5** exemplified in their reactions with morpholine.

## Results and Discussion

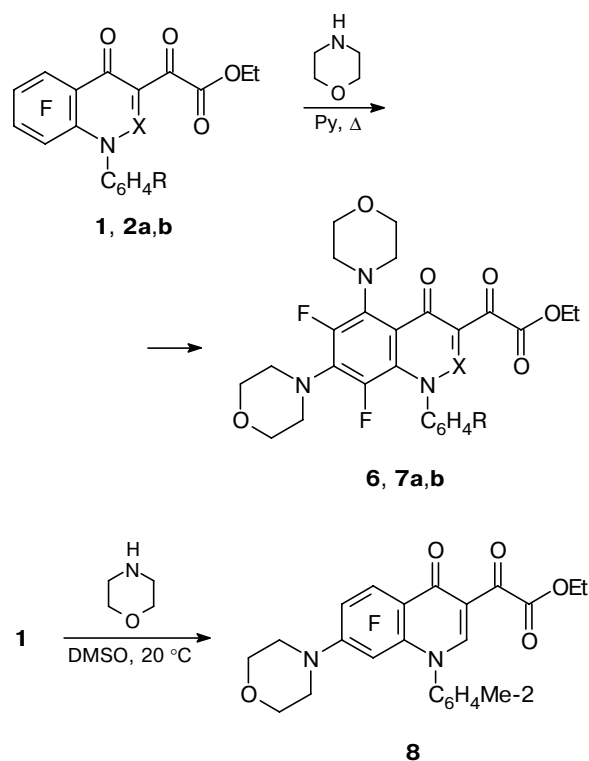
It was found that the reactions of 3-ethoxalyl-quinolone **1** and -cinnolones **2a,b** with an excess of morpholine in boiling pyridine yield 5,7-disubstitution products **6** and **7a,b**, respectively (Scheme 1, Table 1). The positions of the morpholine residues were determined from the coupling constants of the F atoms in <sup>19</sup>F NMR spectra (see Table 1) with consideration of the literature data.<sup>4,5</sup>

Note that cinnolones react more readily, since their conversion is completed within 1 h, whereas the conversion of quinolones takes 3 h (TLC).

The reaction of quinolone **1** in DMSO at ~20 °C affords 7-morpholino derivative **8** (see Scheme 1, Table 1).

Attempts at preparing an individual monosubstitution product from cinnolone **2** by varying reaction conditions (solvent, temperature, and reagent ratio) were unsuccessful.

Scheme 1



X = CH, R = 2-Me (**1**, **6**)

X = N, R = 4-Br (**2a**, **7a**), 2-Me (**2b**, **7b**)

When studying substitution in quinolone **3** and cinnolones **4a,b** under analogous conditions, we found that 5,7-disubstitution products **9** and **10** are formed in boiling pyridine, while 7-monosubstitution derivatives **11** and **12**, in DMSO at ~20 °C (Scheme 2, see Table 1). As in the aforementioned case, compound **3** proved to be less reactive than the corresponding cinnolones. More-

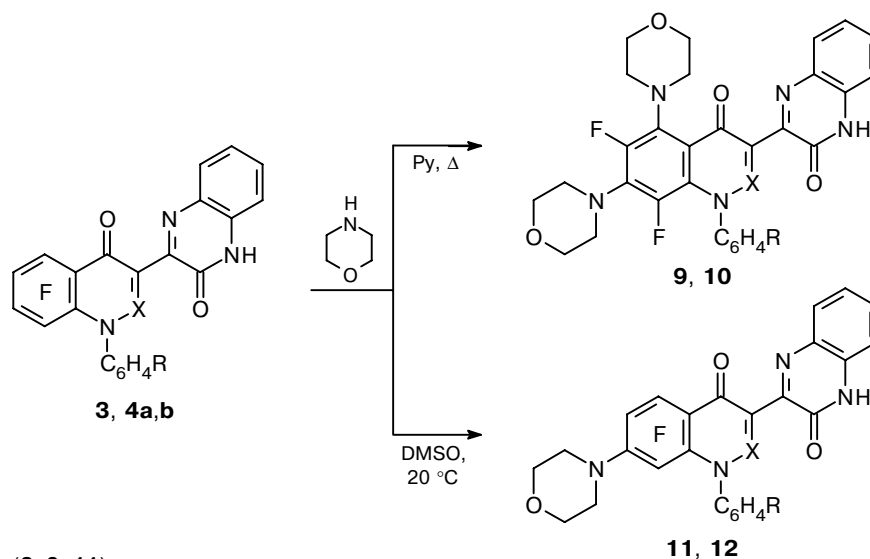
**Table 1.** Main physicochemical parameters of compounds **6–13**

Com- pound	M.p. /°C	Yield (%)	Found (%)				Molecular formula	IR, ν/cm <sup>-1</sup>	NMR (DMSO-d <sub>6</sub> ), δ, J/Hz	
			Calculated						<sup>1</sup> H	<sup>19</sup> F
			C	H	F	N				
<b>6</b>	197—200	75	<u>62.18</u> 62.10	<u>5.36</u> 5.40	<u>6.98</u> 7.02	<u>7.79</u> 7.76	C <sub>28</sub> H <sub>29</sub> F <sub>2</sub> N <sub>3</sub> O <sub>6</sub>	3050 (CH); 1735 (COOEt); 1665 (C=O); 1630 (C=O ring); 1595 (C=N, C=C)	1.33 (t, 3 H, Me, <i>J</i> = 7.1); 2.11 (s, 3 H, Me); 2.96—3.81 (m, 16 H, CH <sub>2</sub> ); 4.33 (q, 2 H, CH <sub>2</sub> , <i>J</i> = 7.1); 7.24—7.57 (m, 4 H, C <sub>6</sub> H <sub>4</sub> ); 8.15 (s, 1 H, CH)	29.57 (br.s, 1 F); 29.83 (br.s, 1 F)
<b>7a</b>	209—210	92	<u>51.70</u> 51.41	<u>4.19</u> 4.15	<u>6.24</u> 6.26	<u>9.03</u> 9.22	C <sub>26</sub> H <sub>25</sub> BrF <sub>2</sub> N <sub>4</sub> O <sub>6</sub>	1725 (COOEt); 1700 (C=O); 1620 (C=O ring); 1595 (C=N, C=C)	1.26 (t, 3 H, Me, <i>J</i> = 7.3); 3.10—3.86 (m, 16 H, CH <sub>2</sub> ); 4.32 (q, 2 H, CH <sub>2</sub> , <i>J</i> = 7.3); 7.65 (m, 4 H, C <sub>6</sub> H <sub>4</sub> )	31.64 (d, 1 F); 32.53 (d, 1 F); <i>J</i> = 6.5
<b>7b</b>	169—171	85	<u>59.70</u> 59.77	<u>5.11</u> 5.20	<u>6.76</u> 7.00	<u>10.30</u> 10.33	C <sub>27</sub> H <sub>28</sub> F <sub>2</sub> N <sub>4</sub> O <sub>6</sub>	1730 (COOEt); 1710 (C=O); 1625 (C=O ring); 1595 (C=N, C=C)	1.27 (t, 3 H, Me, <i>J</i> = 7.3); 3.10—3.86 (m, 16 H, CH <sub>2</sub> ); 4.31 (q, 2 H, CH <sub>2</sub> , <i>J</i> = 7.3); 7.66 (m, 4 H, C <sub>6</sub> H <sub>4</sub> )	31.64 (d, 1 F); 32.53 (d, 1 F); <i>J</i> = 6.4
<b>8</b>	226—228	69	<u>60.79</u> 60.76	<u>4.51</u> 4.46	<u>12.12</u> 12.01	<u>5.93</u> 5.91	C <sub>24</sub> H <sub>21</sub> F <sub>3</sub> N <sub>2</sub> O <sub>5</sub>	3045 (CH); 1735 (COOEt); 1665 (C=O); 1640, 1620 (C=O ring, C=N); 1585 (C=C)	1.31 (t, 3 H, Me, <i>J</i> = 7.1); 2.14 (s, 3 H, Me); 3.03—3.85 (m, 8 H, CH <sub>2</sub> ); 4.33 (q, 2 H, CH <sub>2</sub> , <i>J</i> = 7.1); 7.26—7.58 (m, 4 H, C <sub>6</sub> H <sub>4</sub> ); 8.17 (s, 1 H, CH)	14.74 (dd, 1 F, F(6)); 18.24 (dd, 1 F, F(5)); 26.92 (dd, 1 F, F(8)); <i>J</i> <sub>5,6</sub> = <i>J</i> <sub>6,5</sub> = 19.5; <i>J</i> <sub>5,8</sub> = <i>J</i> <sub>8,5</sub> = 12.2; <i>J</i> <sub>6,8</sub> = <i>J</i> <sub>8,6</sub> = 5.4
<b>9</b>	318—310	89	<u>65.33</u> 65.63	<u>5.07</u> 4.99	<u>6.42</u> 6.49	<u>11.72</u> 11.96	C <sub>32</sub> H <sub>29</sub> F <sub>2</sub> N <sub>5</sub> O <sub>4</sub>	1660 (C=O lactam); 1625 (C=O); 1610, 1595 (C=N, C=C)	2.22 (s, 3 H, Me); 3.05—3.71 (m, 16 H, CH <sub>2</sub> ); 7.44—7.71 (m, 8 H, 2 C <sub>6</sub> H <sub>4</sub> ); 7.89 (s, 1 H, CH); 12.28 (br.s, 1 H, NH)	29.76 (s, 1 F); 29.86 (s, 1 F)
<b>10</b>	>300	96	<u>61.69</u> 61.79	<u>4.73</u> 4.68	<u>6.10</u> 6.31	<u>13.77</u> 13.95	C <sub>31</sub> H <sub>28</sub> F <sub>2</sub> N <sub>6</sub> O <sub>5</sub>	1770 (C=O lactam); 1620 (C=O); 1595 (C=N, C=C)	3.09—3.84 (m, 16 H, CH <sub>2</sub> ); 3.79 (s, 3 H, OMe); 6.96—7.85 (m, 8 H, C <sub>6</sub> H <sub>4</sub> ); 12.60 (s, 1 H, NH)	29.81 (br.s, 1 F); 29.36 (br.s, 1 F)
<b>11</b>	328—330	88	<u>64.89</u> 64.86	<u>4.05</u> 4.08	<u>10.70</u> 10.99	<u>10.89</u> 10.81	C <sub>28</sub> H <sub>21</sub> F <sub>3</sub> N <sub>4</sub> O <sub>3</sub>	3150 (NH); 1685 (C=O lactam); 1620 (C=O); 1590 (C=C, C=N)	2.21 (s, 3 H, Me) 3.03—3.71 (m, 8 H, CH <sub>2</sub> ); 7.21—7.64 (m, 8 H, 2 C <sub>6</sub> H <sub>4</sub> ); 7.75 (s, 1 H, CH) 12.28 (br.s, 1 H, NH)	11.42 (d, 1 F, F(6)); 17.11 (dd, 1 F, F(5)); 25.25 (d, 1 F, F(8)); <i>J</i> <sub>5,6</sub> = <i>J</i> <sub>6,5</sub> = 18.8; <i>J</i> <sub>5,8</sub> = <i>J</i> <sub>8,5</sub> = 13.3

(to be continued)

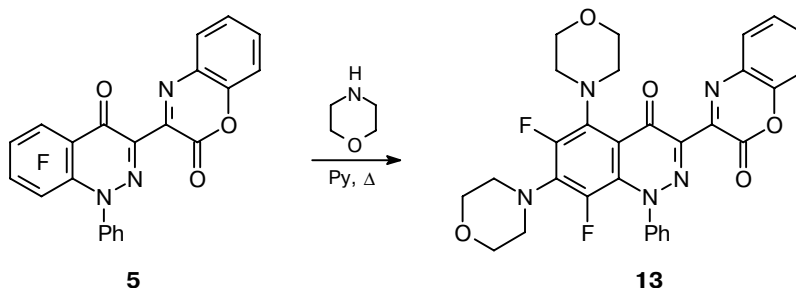
**Table 1** (*continued*)

Compound	M.p. /°C	Yield (%)	Found (%)				Molecular formula	IR, $\nu/\text{cm}^{-1}$	NMR (DMSO- $d_6$ ), $\delta$ , J/Hz	
			C	H	F	N			$^1\text{H}$	$^{19}\text{F}$
<b>12</b>	>300	54	61.52 61.73	3.42 3.59	11.46 11.28	13.53 13.86	$\text{C}_{26}\text{H}_{18}\text{F}_3\text{N}_5\text{O}_3$	3460 (NH); 1665 (C=O lactam); 1630 (C=O); 1585 (C=C, C=N)	3.24, 3.67 (both m, 8 H each, $\text{CH}_2$ ); 7.33–7.84 (m, 9 H, $\text{C}_6\text{H}_4$ , $\text{C}_6\text{H}_5$ ); 12.67 (s, 1 H, NH)	13.89 (dd, 1 F, F(6)); 16.58 (dd, 1 F, F(5)); 32.62 (dd, 1 F, F(8)); $J_{5,6} =$ $J_{6,5} = 19.4$ ; $J_{5,8} = J_{8,5} = 13.6$ ; $J_{6,8} = J_{8,6} = 5.7$
<b>13</b>	278–280	50	62.87 62.82	4.32 4.39	6.70 6.63	12.22 12.21	$\text{C}_{30}\text{H}_{25}\text{F}_2\text{N}_5\text{O}_5$	1750 (C=O lactone); 1625 (C=O); 1590 (C=N, C=C)	3.04–3.29 (m, 16 H, $\text{CH}_2$ ); 6.77–7.91 (m, 9 H, $\text{C}_6\text{H}_4$ , $\text{C}_6\text{H}_5$ )	30.28 (br.s, 1 F); 30.77 (br.s, 1 F)

**Scheme 2**X = CH, R = 2-Me (**3**, **9**, **11**)X = N, R = 4-OMe (**4a**, **10**), H (**4b**, **12**)

over, in the case of quinoxalinonylquinolone **3**, the reaction in DMSO is completed only in the presence of  $\text{Et}_3\text{N}$ .

Cinnolinonylbenzooxazinone **5** reacts with an excess of morpholine in pyridine to give 5,7-disubstitution product **13** (Scheme 3, see Table 1). In DMSO, the

**Scheme 3**

benzooxazinone ring of compound **5** undergoes opening, yielding morpholide and an unseparable mixture of substitution products for F atoms. The same result was obtained in attempting to replace the F atoms in tetrafluoroquinolinonylbenzooxazinone.

Thus, one can conclude that morpholine predominantly attacks position 7 of the cinnolone (quinolone) system. In addition, our results suggest that aromatic nucleophilic substitution in the case of quinolone **1** is slower and more selective than  $S_NAr$  reactions of its aza analogs, cinnolones **2**.

### Experimental

IR spectra were recorded on a Specord IR-75 spectrometer (400–4000  $\text{cm}^{-1}$ , Vaseline oil).  $^1\text{H}$  NMR spectra were recorded on Tesla BS-587 A (80 MHz) and Bruker DRX-400 (400 MHz) spectrometers relative to  $\text{Me}_4\text{Si}$ ;  $^{19}\text{F}$  NMR spectra were recorded on a Tesla BS-587 A spectrometer (75 MHz) relative to  $\text{C}_6\text{F}_6$ . Elemental analysis was carried out on a Carlo Erba CHNS-O EA 1108 instrument.

The starting 3-ethoxalyl-5,6,7,8-tetrafluoro-1-(2-tolyl)-1,4-dihydroquinolin-4-one (**1**),<sup>1</sup> 1-aryl-3-ethoxalyl-5,6,7,8-tetrafluoro-1,4-dihydrocinnolin-4-ones **2a,b**,<sup>2</sup> and 1-aryl-5,6,7,8-tetrafluoro-3-heteryl-1,4-dihydroquinolin(cinnolin)-4-ones **3**, **4a,b**, and **5** (see Ref. 3) were prepared according to the known procedures.

**3-Ethoxalyl-6,8-difluoro-5,7-dimorpholino-1-(2-tolyl)-1,4-dihydroquinolin-4-one (6).** Morpholine (0.43 g, 5 mmol) was added to a solution of quinolone **1** (0.41 g, 1 mmol) in 30 mL of dry pyridine. The reaction mixture was refluxed for 3 h and concentrated. The residue was dissolved in 50 mL of  $\text{CHCl}_3$  and washed with 5% HCl (100 mL) and water to pH 7. The chloroform layer was separated and dried with  $\text{MgSO}_4$ . The solvent was removed, and the residue was recrystallized from MeOH to give product **6** (0.38 g) (see Table 1).

**1-(4-Bromophenyl)-3-ethoxalyl-6,8-difluoro-5,7-dimorpholino-1,4-dihydrocinnolin-4-one (7a).** Analogously, product **7a** (0.9 g) was obtained from cinnolone **2a** (0.85 g, 1.8 mmol) and morpholine (0.8 g, 9 mmol) over 30 min (see Table 1).

**3-Ethoxalyl-6,8-difluoro-5,7-dimorpholino-1-(2-tolyl)-1,4-dihydrocinnolin-4-one (7b).** Analogously, product **7b** (1.38 g) was obtained from cinnolone **2b** (1.18 g, 3 mmol) and morpholine (1.31 g, 9 mmol) (see Table 1).

**3-Ethoxalyl-5,6,8-trifluoro-7-morpholino-1-(2-tolyl)-1,4-dihydroquinolin-4-one (8).** Morpholine (0.43 g, 5 mmol) was added to a solution of quinolone **1** (0.407 g, 1 mmol) in 30 mL of dry DMSO. The reaction mixture was kept at 20 °C for 4 days and poured into 50 mL of 5% HCl. The precipitate that formed was filtered off, washed with water, and dried. Recrystallization from  $\text{Pr}^i\text{OH}$  gave product **8** (0.31 g) (see Table 1).

**3-[6,8-Difluoro-5,7-dimorpholino-4-oxo-1-(2-tolyl)-1,4-dihydroquinolin-3-yl]-1,2-dihydroquinoxalin-2-one (9).** Morpholine (1.37 mL, 10 mmol) was added to a solution of compound **3** (0.7 g, 1.6 mmol) in 20 mL of pyridine. The reaction mixture was refluxed for 10 h and concentrated, and the residue was dissolved in 70 mL of  $\text{CHCl}_3$ . The chloroform

solution was washed with 5% HCl (100 mL) and water to pH 7, dried with  $\text{MgSO}_4$ , and concentrated. Recrystallization of the residue from  $\text{Pr}^i\text{OH}$  gave compound **9** (0.83 g) (see Table 1).

**6,8-Difluoro-1-(4-methoxyphenyl)-5,7-dimorpholino-3-(2-oxo-1,2-dihydroquinoxalin-3-yl)-1,4-dihydrocinnolin-4-one (10).** Analogously, product **10** (0.14 g) was obtained from compound **4a** (0.35 g, 0.8 mmol) and morpholine (0.645 g, 7.4 mmol) over 1 h (see Table 1).

**3-[5,6,8-Trifluoro-7-morpholino-4-oxo-1-(2-tolyl)-1,4-dihydroquinolin-3-yl]-1,2-dihydroquinoxalin-2-one (11).** Triethylamine (0.28 mL, 2 mmol) and morpholine (0.09 g, 1 mmol) were added to a solution of compound **3** (0.45 g, 1 mmol) in 10 mL of DMSO. The reaction mixture was kept at 20 °C for 190 h and poured into 100 mL of 5% HCl. The precipitate that formed was filtered off, washed with water, and recrystallized from  $\text{Pr}^i\text{OH}$ . The yield of compound **11** was 0.46 g (see Table 1).

**5,6,8-Trifluoro-7-morpholino-3-(2-oxo-1,4-dihydroquinoxalin-3-yl)-1-phenyl-1,4-dihydrocinnolin-4-one (12).** Morpholine (0.17 g, 1.5 mmol) was added to a solution of compound **4b** (0.18 g, 0.38 mmol) in 30 mL of DMSO. The reaction mixture was kept at 20 °C for 90 h and poured into 100 mL of 5% HCl. The precipitate that formed was filtered off, washed with water, and recrystallized from  $\text{Pr}^i\text{OH}$ . The yield of compound **12** was 0.11 g (see Table 1).

**3-[6,8-Difluoro-5,7-dimorpholino-4-oxo-1-phenyl-1,4-dihydrocinnolin-3-yl]-1,2-dihydrobenzooxazin-2-one (13).** Analogously, compound **13** (0.143 g) was obtained from compound **5** (0.23 g, 0.5 mmol) and morpholine (0.44 g, 5 mmol) (see Table 1).

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